

# Lewis acid promoted C–C and copper-catalyzed C–O bond formation: synthesis of neoflavans†

B. Suchand, J. Krishna, K. Mritunjy and G. Satyanarayana\*

Cite this: *RSC Adv.*, 2014, 4, 13941

Received 3rd January 2014  
Accepted 30th January 2014

DOI: 10.1039/c4ra00048j

www.rsc.org/advances

An intramolecular [Cu]-catalyzed C–O bond formation for the synthesis of neoflavans is presented. Lewis acid promoted Friedel–Crafts Michael addition of electron rich aromatic systems onto the double bond of the cinnamate ester was employed to furnish a  $\beta$ -diaryl ester. Electrophilic aromatic bromination of the  $\beta$ -diaryl ester and reduction/Grignard addition furnished the required precursor alcohols. The method is applicable to the synthesis of neoflavans containing tertiary as well as quaternary carbon centers. Significantly, the neoflavan substructures are present in biologically active compounds.

## Introduction

The neoflavans (4-aryl-3,4-dihydro-2*H*-chromenes) or substituted chromans are ubiquitous substructures present in biologically active natural compounds.<sup>1–3</sup> For example, the *Dalbergia* plant species are used in traditional Chinese medicine for curing inflammation, blood disorders and ischemia.<sup>4</sup> Similarly, the simple neoflavene dalbergichromene, isolated from the stem bark of *Dalbergia* species,<sup>5,6</sup> shows medicinal activity. Also, the chroman natural product centchroman exhibits antifertility properties as an estrogen antagonist, which was isolated from the methanolic extracts of *A. indicum*.<sup>7</sup> In addition, 3-hydroxy-4-arylchroman is also observed as a substructure in 4-arylflavan-3-ol, isolated from the South African plant *Nelia meyeri*.<sup>8</sup> Further, the flavonoid myristinin A possessing the 4-arylchroman part structure is a potent polymerase beta inhibitor (Fig. 1).<sup>9</sup>

Owing to the interesting structural features and important biological properties of neoflavans/flavones, there are a number of reports towards their synthesis.<sup>10</sup> Notably, Wang *et al.* reported enantioselective synthesis of chromans and dihydrobenzopyrans by employing Friedel–Crafts alkylation

and cyclization of 1-naphthol and  $\alpha,\beta$ -unsaturated aldehydes.<sup>10c</sup> Benzopyrans were produced by Pettus and co-workers using enantioselective [4 + 2] cycloaddition between *ortho*-quinone methides (*o*-QM) and enol ethers.<sup>10f</sup> The research group of Sames *et al.* developed new method for the construction of the chromene core by means of Pt(IV)-catalyzed intramolecular hydroarylation of arene–alkyne substrates.<sup>10g</sup> The research group of Tunge disclosed a three-step stereoselective reduction as the key step for the synthesis of diarylchromans from phenols and cinnamic acid.<sup>10h</sup> To the best of our knowledge, thus far there is no report on the synthesis of neoflavans possessing a quaternary C-4 carbon atom.

In continuation of our research interest in transition-metal catalysis,<sup>11</sup> recently we have disclosed an efficient three-step strategy for the synthesis of flavans and benzoxepines by employing [Pd]-catalyzed C–C and [Cu]-catalyzed C–O

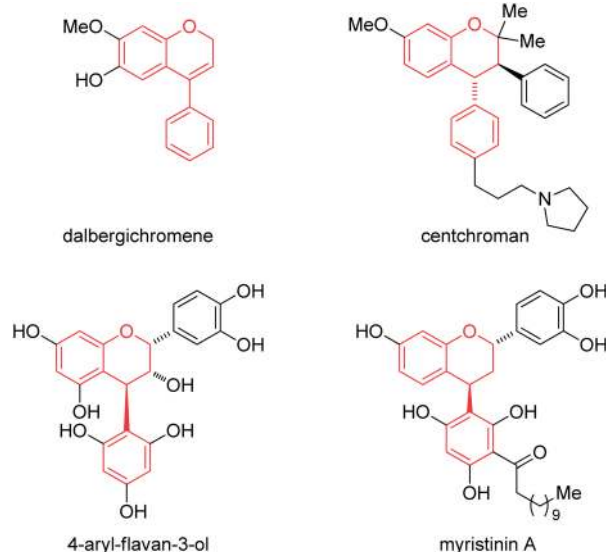
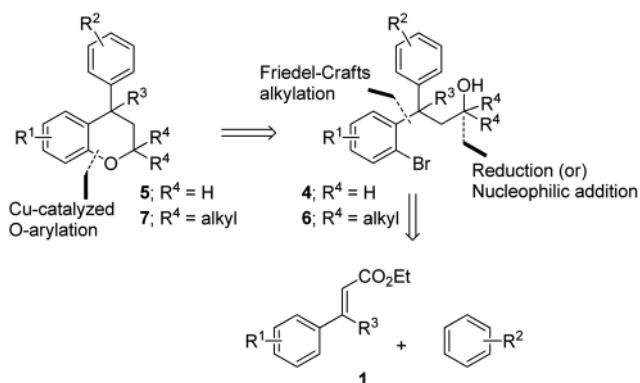


Fig. 1 Representative examples of naturally occurring neoflavan natural products containing neoflavan scaffolds.

Department of Chemistry, Indian Institute of Technology (IIT) Hyderabad, Ordnance Factory Estate Campus, Yeddumailaram - 502 205, Medak District, Andhra Pradesh, India. E-mail: gvsatya@iith.ac.in; Fax: +91 40 2301 6032; Tel: +91 40 2301 6054

† Electronic supplementary information (ESI) available: See DOI: 10.1039/C4RA00048J



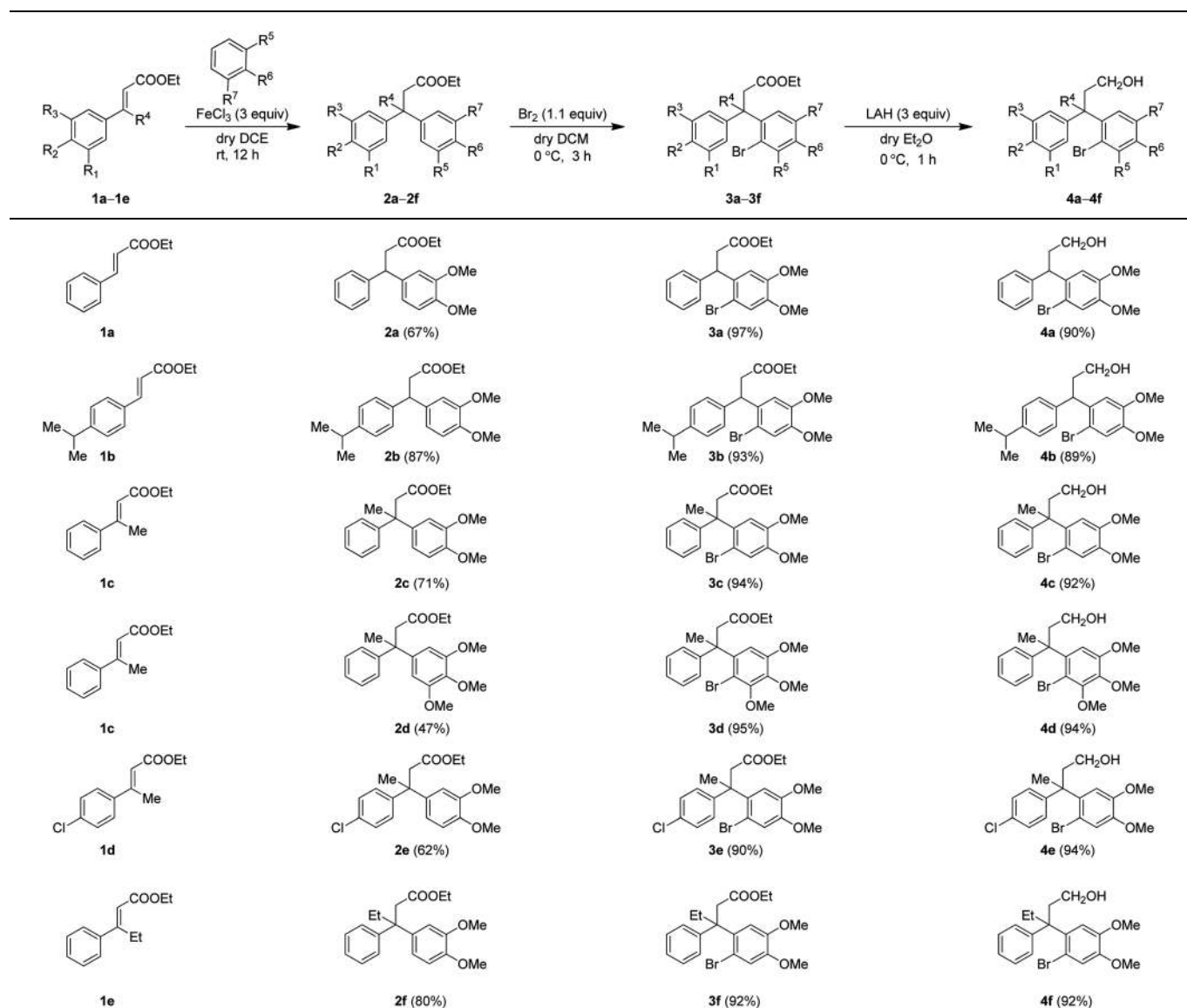


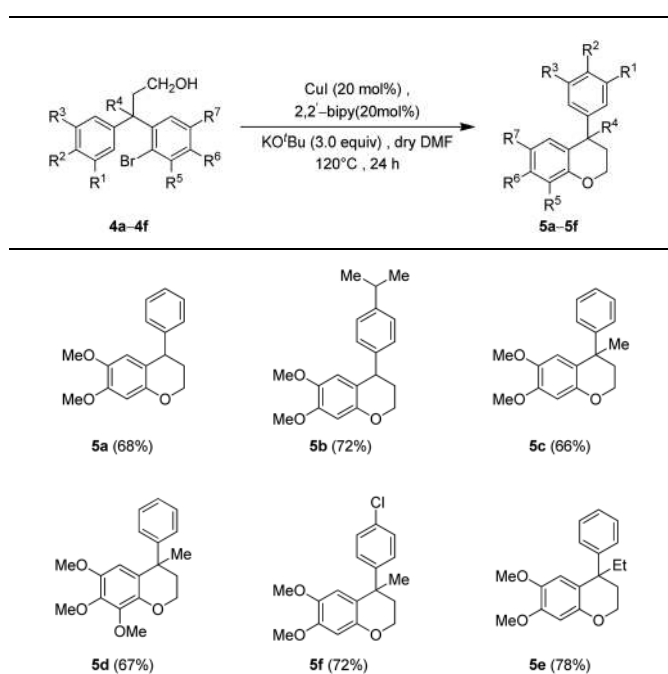
Scheme 1 Retrosynthetic plan for flavans 5 and 7 from cinnamates 1.

bond-forming reactions as the key steps.<sup>12</sup> Herein, we present a practical method for the synthesis of neoflavans, wherein Lewis acid (FeCl<sub>3</sub>) promoted Friedel–Crafts Michael addition and an intramolecular [Cu]-catalyzed Buchwald–Hartwig coupling reaction are the key transformations involved in the strategy.

## Results and discussion

We envisioned that the substituted neoflavans 5 and 7 can be obtained by a key [Cu]-mediated intramolecular Buchwald–Hartwig C–O bond formation between aryl bromide and hydroxy (primary/tertiary) functionality of 4 and 6 respectively. The required primary/tertiary alcohols 4 and 6 were produced from the corresponding esters, which were in turn accessed by

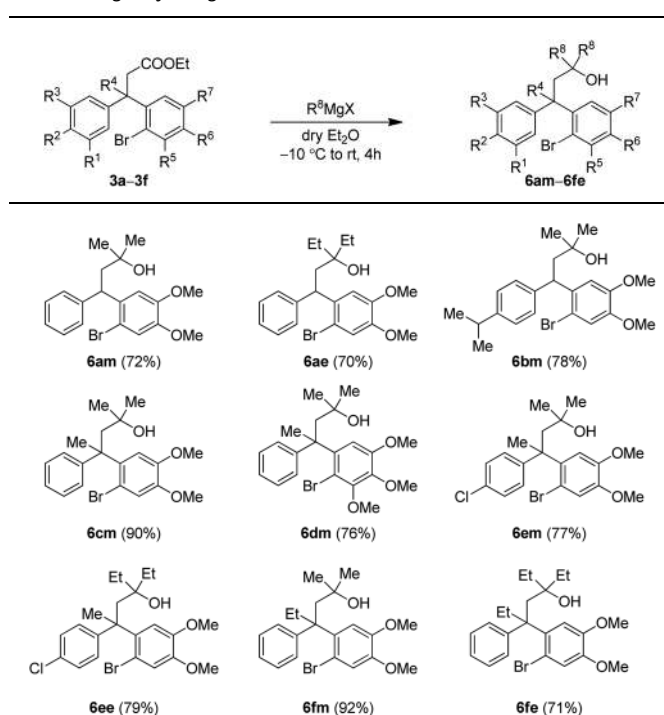
Table 1 Synthesis of primary alcohols 4a–4f starting from cinnamates 1a–1e via β-diaryl esters 2a–2f and bromo-esters 3a–3f<sup>a</sup><sup>a</sup> Isolated yields of chromatographically pure products.

**Table 2** [Cu]-catalyzed synthesis of neoflavans **5a–5f** from primary alcohols **4a–4f**<sup>a,b</sup>

<sup>a</sup> Isolated yields of chromatographically pure products. <sup>b</sup> Reaction was carried out on alcohols **4a–4f** (0.30 mmol) in DMF (3 mL).

employing Lewis acid (FeCl<sub>3</sub>) induced controlled Friedel–Crafts alkylation (Michael addition type) of an electron rich external arene on ethyl cinnamates **1** followed by preferential electrophilic bromination of the electron rich aromatic ring particularly at the *ortho*-position to the ester tether (Scheme 1). It is worth mentioning that the controlled Friedel–Crafts alkylation was observed, particularly, by the Lewis acid (FeCl<sub>3</sub>) in our previous study, which formed the basic foundation for this work.<sup>13</sup>

The synthetic study began with the controlled Friedel–Crafts alkylation (*i.e.* the reaction impeded after Friedel–Crafts Michael addition without allowing it to proceed to subsequent intramolecular acylation) of ethyl cinnamates **1** with electron rich aromatic systems such as 1,2-dimethoxybenzene or 1,2,3-trimethoxybenzene. Thus, treatment of simple as well as  $\beta$ -alkyl substituted ethyl cinnamates **1** with the external arene in the presence of Lewis acid (FeCl<sub>3</sub>, 3 equiv.), at ambient temperature, furnished the  $\beta$ -arylated esters **2a–2f**, containing tertiary/quaternary carbon atom, in fair to very good yields (Table 1).<sup>13</sup> It is worth mentioning that the use of DCE as a medium was found to be more general than CH<sub>2</sub>Cl<sub>2</sub> in order to give good yields of the products **2**. However, the same Friedel–Crafts Michael addition either with external arene on 2-bromocinnamate or with external bromoarene on simple cinnamate failed to furnish the corresponding product.<sup>14</sup> Preferential bromination of the electron rich aromatic ring under the standard electrophilic bromination conditions, provided the esters **3a–3f** in excellent yields as shown in Table 1. It is worth mentioning that when both the aromatic rings were electron rich, the

**Table 3** Synthesis of tertiary alcohols **6am–6fe** from bromo-esters **3a–3f** using alkyl Grignard reaction<sup>a,b,c</sup>

<sup>a</sup> Isolated yields of chromatographically pure products. <sup>b</sup> The first letter (**a–f**) for compounds **6** indicates the starting material, while the second letter (**m** or **e**) refers to methyl or ethyl groups, respectively. <sup>c</sup> 8 equiv. of alkyl Grignard reagent was used.

bromination was not selective and gave a mixture of brominated products. Next, reduction of the ester function was required to provide the corresponding alcohols for the planned metal mediated cyclization. The esters **3a–3f** were reduced with LiAlH<sub>4</sub>, and the precursor primary alcohols **4a–4f** yielded in near quantitative yields, in a controlled manner without affecting the bromo-substituent (Table 1).

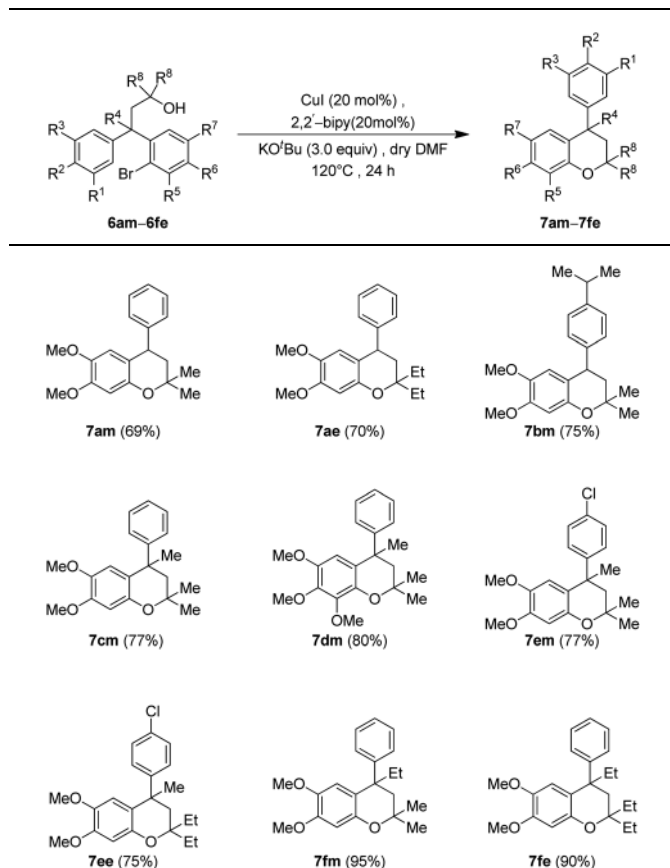
With the precursor alcohols **4** in hand, the key transition metal catalyzed intramolecular C–O bond formation was explored. In this regard, we employed the same reaction conditions [CuI (20 mol%)/2,2'-bipyridyl (20 mol%), base KOtBu (3 equiv.) in hot DMF (120 °C) for 24 h] which were successful in our previous reports for the synthesis of flavans.<sup>12</sup> As expected, the reaction was quite successful and furnished the neoflavan **5a** in very good yield.

Since the above [Cu]-catalyzed C–O bond forming reaction conditions were successful for the construction of neoflavan **5a** (Table 2), these conditions were applied to the other alcohols **4b–4f** as well. Agreeably, these conditions were found to be versatile and gave the cyclized neoflavans **5b–5f** (Table 2) possessing tertiary and quaternary C-4 carbon centers, in very good yields.

After successful synthesis of various neoflavan derivatives **5a–5f** with varied functionalities on either aromatic ring starting from primary alcohol precursors **4a–4f**, to check the scope and limitations of the method, we turned our attention to the



Table 4 [Cu]-catalyzed synthesis of neoflavans **7am–7fe** from tertiary alcohols **6am–6fe**<sup>a,b</sup>



<sup>a</sup> Isolated yields of chromatographically pure products. <sup>b</sup> Reaction was carried out on alcohols **6am–6fe** (0.30 mmol) in DMF (3 mL).

synthesis of tertiary alcohols. Thus, the reaction of  $\beta$ -diaryl esters **3** with Grignard reagents furnished the corresponding tertiary alcohols **6** as described in Table 3.

Finally, the key intramolecular C–O bond formation catalyzed by a copper metal complex was explored. As expected, the method was found to be quite successful on the tertiary alcohols as well and generated the cyclized neoflavans, **7am–7fe**, in very good yields (Table 4).

## Conclusions

In summary, we have developed a simple and practically applicable four-step strategy for the synthesis of functionalized neoflavans. Intermolecular Friedel–Crafts alkylation (C–C bond formation) and intramolecular [Cu]-catalyzed C–O bond formation were applied as key steps of the strategy. The strategy is efficient and amenable to the synthesis of neoflavans containing tertiary as well quaternary carbon centers.

## Acknowledgements

Financial support by the Department of Science and Technology [(DST), CHE/2010-11/006/DST/GSN], New Delhi is gratefully

acknowledged. J. K. thanks CSIR, New Delhi, for the award of a research fellowship.

## Notes and references

- W. B. Eyton, W. D. Ollis, I. O. Sutherland, O. R. Gottlieb, M. T. Magalhaes and L. M. Jackman, *Tetrahedron*, 1965, **21**, 2683.
- D. M. X. Donnelly and G. Boland, *The Flavonoids: Advances in Research since 1986*, ed. J. B. Harborne, Chapman & Hall, London, 1994, p. 239.
- (a) D. A. Horton, G. T. Bourne and M. L. Smythe, *Chem. Rev.*, 2003, **103**, 893; (b) H. C. Shen, *Tetrahedron*, 2009, **65**, 3931.
- S. C. Chan, Y. S. Chang and S. C. Kuo, *Phytochemistry*, 1997, **46**, 947.
- S. K. Mukerjee, T. Saroja and T. R. Seshadri, *Tetrahedron*, 1971, **27**, 799.
- V. K. Dhingra, S. K. Mukerjee, T. Saroja and T. R. Seshadri, *Phytochemistry*, 1971, **10**, 2551.
- M. S. Sankaran and M. R. N. Prasad, *Contraception*, 1974, **9**, 279.
- H. Kolodziej, *Phytochemistry*, 1984, **23**, 1745.
- (a) J.-Z. Deng, S. R. Starck, S. Li and S. M. Hecht, *J. Nat. Prod.*, 2005, **68**, 1625; (b) D. J. Maloney, J.-Z. Deng, S. R. Starck, Z. Gao and S. M. Hecht, *J. Am. Chem. Soc.*, 2005, **127**, 4140; (c) S. Sawadjoon, P. Kittakoop, K. Kirtikara, V. Vichai and M. Y. Tanticharoen, *J. Org. Chem.*, 2002, **67**, 5470.
- (a) D. M. X. Donnelly, J.-P. Finet, P. J. Guiry and K. Nesbitt, *Tetrahedron*, 2001, **57**, 413; (b) S.-R. Li, L.-Y. Chen, J.-C. Tsai, J.-Y. Tzeng, I.-L. Tsai and E.-C. Wang, *Tetrahedron Lett.*, 2007, **48**, 2139; (c) L. Hong, L. Wang, W. Sun, K. Wong and R. Wang, *J. Org. Chem.*, 2009, **74**, 6881; (d) J. Barluenga, M. Trincado, E. Rubio and J. M. González, *J. Am. Chem. Soc.*, 2004, **126**, 3416; (e) W. Quaglia, M. Pignini, A. Piergentili, M. Giannella, F. Gentili, G. Marucci, A. Carrieri, A. Carotti, E. Poggesi, A. Leonardi and C. Melchiorre, *J. Med. Chem.*, 2002, **45**, 1633; (f) C. Selenski and T. R. Pettus, *J. Org. Chem.*, 2004, **69**, 9196; (g) S. J. Pastine, S. W. Youn and D. Sames, *Org. Lett.*, 2003, **5**, 1055; (h) K. Li, K. Vanka, W. H. Thompson and J. A. Tunge, *Org. Lett.*, 2006, **8**, 4711; (i) B. D. Gallagher, B. R. Taft and B. H. Lipshutz, *Org. Lett.*, 2009, **11**, 5374; (j) J. D. Chambers, J. Crawford, H. W. R. Williams, C. Dufresne, J. Scheiget, M. A. Bernstein and C. K. Lau, *Can. J. Chem.*, 1992, **70**, 1717; (k) G. Liu and X. Lu, *Tetrahedron*, 2008, **64**, 7324; (l) A. Volonterio and M. Zanda, *Tetrahedron Lett.*, 2005, **46**, 8723.
- (a) A. G. K. Reddy, J. Krishna and G. Satyanarayana, *Synlett*, 2011, 1756; (b) J. Krishna, A. G. K. Reddy, L. Mahendar, B. V. Ramulu and G. Satyanarayana, *Synlett*, 2012, **23**, 375; (c) A. G. K. Reddy and G. Satyanarayana, *Tetrahedron*, 2012, **68**, 8003; (d) A. G. K. Reddy, J. Krishna and G. Satyanarayana, *Tetrahedron Lett.*, 2012, **53**, 5635; (e) L. Mahendar, J. Krishna, A. G. K. Reddy, B. V. Ramulu and G. Satyanarayana, *Org. Lett.*, 2012, **14**, 628; (f) J. Krishna, A. G. K. Reddy and G. Satyanarayana, *Synlett*, 2013, **24**, 967; (g) A. G. K. Reddy, J. Krishna and G. Satyanarayana, *Tetrahedron*, 2013, **69**, 10098; (h) J. Krishna, A. G. K. Reddy and G. Satyanarayana, *Tetrahedron Lett.*, 2014, **55**, 861.



## Communication

- 12 (a) B. Suchand, J. Krishna, B. V. Ramulu, D. Dibyendu, A. G. K. Reddy, L. Mahendar and G. Satyanarayana, *Tetrahedron Lett.*, 2012, 53, 3861; (b) B. V. Ramulu, L. Mahendar, J. Krishna, A. G. K. Reddy, B. Suchand and G. Satyanarayana, *Tetrahedron*, 2013, 69, 8305.
- 13 B. V. Ramulu, A. G. K. Reddy and G. Satyanarayana, *Synlett*, 2013, 24, 868.

14

